# Clayton Valley Charter High School 

Chemistry

## Benchmark \#4

Review

## 2017-2018

Name: $\qquad$ Period: $\qquad$

Student Instructions:

1. Complete the entire review guide
2. This will count for credit.
3. Anything that you struggle with is something that you need to study.
4. Do not procrastinate - you will only struggle that much more on the benchmark exam.
5. YES, the benchmark counts for actual points in your grade. Not extra credit.

## Section 1: Gas Laws and Kinetic Molecular Theory

1. A Peep is taken from the cupboard and put in the microwave. If the cupboard is at $25^{\circ} \mathrm{C}$ and the Peep's initial volume is 100 mL , what will the volume of the Peep be if it is heated to $75^{\circ} \mathrm{C}$ ?
2. A deep sea diver seals a 20 mL water bottle at sea level, where the pressure is 1 atm . He takes it on a dive to 100 ft below sea level, where the pressure is 4 atm. What will be the volume of the bottle at this depth?
3. "Deflategate" was a controversial issue in the NFL, where the New England Patriots were accused of deflating balls during a playoff game to gain an unfair advantage. The rules state that balls must be inflated to a minimum of 12.5 psi in a locker room environment at $22^{\circ} \mathrm{C}$. At kickoff, the temperature measured $9^{\circ} \mathrm{C}$.
a. Assuming the balls were filled at exactly 12.5 psi before the game, what should the halftime pressure of the balls be?
b. The balls were found to be at 11.1 psi at halftime. Using this data, what would have the original psi of the balls been, assuming same temperature conditions?
4. A 4.5 L container of gas has a pressure of 3 atm at $100^{\circ} \mathrm{C}$. The container is expanded to 6.0 L and the temperature is increased to $200^{\circ} \mathrm{C}$. What is the final pressure of the container?

Use the graph to answer the following questions.

5. Define the two types of graphical relationships we've discussed:
a. Direct Relationship:
b. Indirect/Inverse Relationship:
c. [Circle One] Does the graph depict an INDIRECT or DIRECT relationship?
d. Describe the movement of the particles at the point $(0,0)$.
6. Qatar is hosting the World Cup in 2022. The average temperatures in June in Qatar are $42^{\circ} \mathrm{C}$. Convert this to ${ }^{\circ} \mathrm{F}$ and Kelvin.
7. Propose what happens to particles in a system (whether a container, a room or the universe) if absolute zero is reached. Support this explanation by drawing a particle model.


For numbers $9-10$, refer to the figure above.
8. Explain two different ways of how to make the pressure gauge increase.
9. Explain two different ways how to make the temperature decrease.
10. Snakes have evolved to use infrared radiation in their vision to capture their prey. Explain the long-term effects of global warming and its potential effects on snakes.
11. Compare and contrast the Greenhouse Effect by drawing a diagram of a car on a hot day and the Earth's atmosphere.

