

UNIT 3: ATOMS, ELEMENTS, MOLECULES

Lesson 14: Holey Moley

Scientific Notation



GUIDING QUESTION: HOW CAN YOU TRANSLATE NUMBERS INTO SCIENTIFIC NOTATION? EXPLAIN HOW TO CALCULATE THE MOLAR MASS OF AN ELEMENT.

Do Now:

If you have 2 moles of snowflakes, how many snowflakes do you have (assuming none melt)?



WHAT IS A MOLE?



MOLES:

How many candies are in a mole of candies?

How many molecules are in a mole of water, H₂O? How many atoms are there in a mole of sodium?

Why do scientists use moles?

One mole of hydrogen atoms has a mass of 1.01 grams. How many atoms would be in 3.03 grams of hydrogen? How do you know?

One mole of oxygen atoms has a mass of 15.99 grams. How many atoms would be in 31.98 grams of oxygen? How do you know?