UNIT 3: ATOMS, ELEMENTS, MOLECULES

## Scientific Notation <br>  <br> $4,007,500,000 \mathrm{~cm} .00000001 \mathrm{~cm}$

GUIDING QUESTION: HOW CAN YOU TRANSLATE NUMBERS INTO SCIENTIFIC NOTATION? EXPLAIN HOW TO CALCULATE THE MOLAR MASS OF AN ELEMENT.

## Do Now:

If you have 2 moles of snowflakes, how many snowflakes do you have (assuming none melt)?


## WHAT IS A MOLE?



## MOLES:

How many candies are in a mole of candies?
How many molecules are in a mole of water, $\mathrm{H}_{2} \mathrm{O}$ ? How many atoms are there in a mole of sodium?

Why do scientists use moles?
One mole of hydrogen atoms has a mass of 1.01 grams. How many atoms would be in 3.03 grams of hydrogen? How do you know?

One mole of oxygen atoms has a mass of 15.99 grams. How many atoms would be in 31.98 grams of oxygen? How do you know?

